## A New Approach to the Synthesis of the Strychnos Alkaloids Core Structure

## Philip Magnus\* and Melvyn Giles

Department of Chemistry and Biochemistry, University of Texas at Austin, Austin, Texas 78712

Abstract: The nine-membered ring sec-amine 4 reacts with methyl propiolate/sodium hydride in toluene at reflux to give the strychnos core structure 12 in a single step.

We recently reported the synthesis of strychnine 1 via the relay compound 7.1 The overall strategy, outlined in Scheme 1, involved the chloroformate induced cleavage of the tetracyclic amine 2 to give the nine-membered ring carbamate 3, and deprotection to give 4 followed by several steps to give the core skeleton 6. The indicated C-C bonds in compound 6 are made in separate steps. The synthesis was completed by conversion of 6 into 7, which was also available by degradation of strychnine. Finally, 7 was converted into strychnine via the Wieland-Gumlich aldehyde. While this work was in progress Kuehne reported that the same nine-membered sec-amine 4 on treatment with n-butanal formed the enamine 5 which at room temperature underwent intramolecular [2+4] cycloaddition to the indole 2-acrylate to give the strychnos alkaloid tubotaiwine.

This offers an extremely rapid and elegant entry into the strychnos skeleton and we were interested to try the Kuehne strategy to simplify the synthesis of

strychnine. Treatment of the sec-amine 4 with acetaldehyde gave a complex mixture of products that did not contain any of the required  $\beta$ -amino acrylate adduct 9 [characteristic UV (MeOH) [ $\lambda_{max}(\epsilon)$ ] 323.3 (12,900)]. The use of acetaldehyde equivalents such as ethyl vinyl ether, vinyl acetate, vinyl imidazole and N-vinyl pyrrolidine did not improve the situation. Consequently, it was decided to examine the reaction of 4 with electron-deficient acetylenes.

Treatment of 4 with methyl propiolate in dichloromethane at room temperature gave the conjugate addition adduct 10 (90%). Heating the adduct 10 in toluene at reflux in the presence of trimethylsilyl triflate gave the rearranged compound 13 in low yield (structure by X-ray). The same reaction at room temperature gave the intermediate 11 (92%). Presumably 13 is formed by further reaction of 11 to give 12, which can  $\beta$ -eliminate to the 1,4-cyclohexadiene 12a and add the liberated  $\sec$ -amine functionality to the  $\beta$ -aminoacrylate, resulting in 13.

Fortunately, we discovered that treatment of 10 with sodium hydride in toluene heated at reflux gave the required adduct 12 (61%). The success of these reaction conditions may be attributed to the formation of 12 as its sodium salt, effectively prohibiting further transformation into 13. A similar sequence of reactions was attempted using phenylethnylsulfone but the initial conjugate addition

adduct (cf. 10) did not react further, consequently we had to distinquish between the two ester groups in 12.

Reduction of the  $\beta$ -aminoacrylate double bond with Zn/H<sub>2</sub>SO<sub>4</sub>/MeOH gave 14, which was epimerized and the indoline nitrogen protected as its p-MeOC<sub>6</sub>H<sub>4</sub>SO<sub>2</sub>-derivative 15 (85%, from 12). Treatment of 15 with LiOH/MeOH/THF (56%) followed by Barton decarboxylation<sup>4</sup> ((DBU/Pr<sup>i</sup>COCl/N-hydroxypyridine-2-thione, hv/Bu'SH) gave 16 (76%). Its structure was confirmed by reduction to the alcohol 17 and comparison with an authentic sample made from the relay compound 7 by silylation, deoxygenation<sup>5</sup> and hydrolysis (16% overall).

We were interested to see if either 15 and/or 16 could be selectively oxidized adjacent to the t-amine and provide a possible handle to attach the hydroxyethylidene group. A number of reagents are available that can oxidize amines to amides via the iminium ion.<sup>6</sup> Treatment of 15 with RuCl<sub>3</sub> (cat)/NaIO<sub>4</sub> gave the five-membered ring amide 18 (50%).<sup>7</sup> Under the same oxidation conditions 16 gave a very complex mixture of products. Attempts to use the alcohol 17 to direct intramolecular functionalization [Pb(OAc)<sub>4</sub>/I<sub>2</sub> etc] resulted in complex mixtures with no evidence for any functionalization either  $\alpha$ - or  $\beta$ - to the t-amine.

## Scheme 5 (R = $SO_2C_6H_4OMe-p$ )

In summary, this strategy offers a very direct route into the strychnos alkaloid skeleton, and may be modified by introducing more functionality into the starting amine 2. The tetracyclic amine 2 is available from a Pictet-Spengler cyclization of tryptamine and dimethyl-2-ketoglutarate followed by reduction of the amide group.8

Acknowledgements. The National Institutes of Health (GM 32718) are thanked for their financial support of this research. Dr. Vince Lynch (UT, Austin) is thanked for the X-ray structure determination of compound 13.

## References and Footnotes.

- 1. Magnus, P.; Giles, M.; Bonnert, R.; Kim, C. S.; McQuire, L.; Merritt, A.; Vicker, N. J. Am. Chem. Soc. 1992, 114, 4403. Woodward, R. B.; Cava, M. P.; Ollis, W. D.; Hunger, A.; Daeniker, H. U.; Schenker, K. J. Am. Chem. Soc. 1954, 76, 4749. Woodward, R. B.; Cava, M. P.; Ollis, W. D.; Hunger, A.; Daeniker, H. U.; Schenker, K. Tetrahedron, 1963, 19, 247. Most recent published synthetic approaches: Fevig, J. M.; Marquis, Jr. R. W.; Overman, L. E. J. Am. Chem. Soc. 1991, 113, 5085. Grotjahn, D. B.; Vollhardt, K. P. C. J. Am. Chem. Soc. 1990, 112, 5653. Kraus, G. A.; Thomas, P. J.; Bougie, D.; Chen, L. J. Org. Chem. 1990, 55, 1624. Legseir, B.; Henin, J.; Massiot, G.; Vercauteren, J. Tetrahedron Lett. 1987, 28, 3573. Bonjoch, J.; Casamitjana, N.; Quirante, J.; Rodriquez, M.; Bosch, J. J. Org. Chem. 1987, 52, 267. Rawal, V. H.; Michoud, C.; Monestel, R. F. J. Am. Chem. Soc. 1993, 115, 3030.
- 2. Wieland, H.; Gumlich, W. Liebigs Ann. Chem. 1932, 494, 191. Wieland, H.; Kaziro, K. Liebigs Ann. Chem. 1933, 506, 60. Deyrup, J. A.; Schmid, H.; Karrer, P. Helv. Chim. Acta. 1962, 45, 2266.
- 3. Kuehne, M. E.; Frasier, D. A.; Spitzer, T. D. J. Org. Chem. 1991, 56, 2696. Apparently the cyclizations reported in this paper are quite specific, and attempts to extend this strategy in a more general sense did not work. Dr. Kuehne is thanked for this information.
- 4. Barton, D. H. R.; Crich, D.; Motherwell, W. B. J. Chem. Soc., Chem. Commun. 1983, 939. Crich, D. Aldrichimica, 1987, 20, 35.
- 5. Barton, D. H. R.; Jaszberenyi, J. Cs. Tetrahedron Letters, 1989, 30, 2619. Barton, D. H. R.; McCombie, S. W. J. Chem. Soc. Perkin Trans I, 1975, 1574.
- 6. Hudlicky, M. "Oxidations in Organic Chemistry", ACS Monograph 186, p. 240. American Chemical Society, Washington, DC, 1990.
- 7. Yoshifujji, S.; Tanaka, K-I.; Kawai, T.; Nitta, Y, Chem. Pharm. Bull. 1986, 34, 3873. Murahashi, S. -I.; Naota, T.; Yonemura, K. J. Am. Chem. Soc. 1988, 110, 8256.
- 8. Magnus, P.; Ladlow, M.; Elliott, J.; Kim, C-S. J. Chem. Soc., Chem. Commun. 1989,
- 518. Wenkert, E.; Garratt, S.; Dave, K. G. Can. J. Chem. 1964, 42, 489. Magnus, P.; Ladlow, M.; Kim, C. S.; Boniface, P. Heterocycles, 1989, 28, 951. Magnus, P.; Stamford, A.; Ladlow, M. J. Am. Chem. Soc. 1990, 112, 8210.